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# Shape Memory Effect Driven by Diffusionless and Diffusional Transformations at Elevated Temperatures

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Several alloy systems can be selected for high-temperature shape-memory alloys, defined as alloys with stable reverse martensitic-transformation temperatures above 100°C. However, due to the lack of minimum quality standards for stability, ductility, functional behaviour, and reliability, no successful applications have been realised so far. Nevertheless, research on high-temperature shapememory alloys (HTSMA) is an important topic not only for scientific reasons but also due to the market pull. This paper reviews some novel HTSMA systems showing shape-memory effect at elevated temperatures driven by martensitic (diffusionless) and diffusional transformations.

Можна навести певну кількість систем сплавів, що можуть бути визначені, як високотемпературні сплави з ефектом пам'яті форми, що в свою чергу визначаються як сплави зі стабільними температурами зворотного мартенситного перетворення, вищими за 100°С. На жаль, невідомо жодного випадку успішного застосування високотемпературних сплавів з ефектом пам'яті форми, завдяки відсутності вимог щодо їх стабільності, пластичності, функціональної поведінки та надійності. Не зважаючи на це, дослідження високотемпературних сплавів з ефектом пам'яті форми є важливим не тільки з точки зору фундаментальної науки, але й завдяки вимогам ринку. Дана робота присвячена розглядові деяких новітніх систем високотемпературних сплавів з ефектом пам'яті форми, що зумовлений мартенситним (бездифузійним) та дифузійним перетвореннями.

Можно привести достаточное количество систем сплавов, которые подпадают под определение высокотемпературных сплавов с эффектом памяти формы — сплавов, в которых стабильное обратное мартенситное превращение протекает при температурах выше 100°С. К сожалению, до сих пор

305

не известно ни одного случая успешного применения высокотемпературных сплавов с эффектом памяти формы в силу отсутствия требований к их стабильности, пластичности, функциональному поведению и надежности. Тем не менее, исследования высокотемпературных сплавов с эффектом памяти формы являются важными не только с точки зрения фундаментальной науки, но и благодаря требованиям рынка. Настоящая работа посвящена рассмотрению некоторых новейших систем высокотемпературных сплавов с эффектом памяти формы, который определяется как мартенситным (бездиффузионным), так и диффузионными превращениями.

**Key words:** shape memory effect, high-temperature martensitic and diffusional transformations.

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# 1. MARTENSITIC TRANSFORMATION AND SHAPE MEMORY EFFECT IN Zr-BASED INTERMETALLIC COMPOUNDS

#### 1.1. Introduction

It is known that the B2 equiatomic intermetallic compounds of Zr (ZrCobased [1], ZrCu [2], ZrRh [3]) undergo a martensitic transformation (MT) with wide temperature hysteresis. The first reports about the results of the investigations of the MT in Zr-based equiatomic intermetallics belong to Wang [4, 5]. It was shown that the high-temperature B2 phase of ZrPd and ZrRh compounds undergo on cooling the MT very close to the case of TiNi. However, the lack of systematic investigations of MT in Zr-based intermetallics (in contrast to TiNi and related compounds) existed until the series of works of Harris with co-authors [1, 6-11]. They have investigated the constitution of ZrCo, ZrFe, ZrNi,  $ZrCo_{(1-x)}Ni_x$ , ZrCu alloys, their mechanical properties [7, 8], crystal structure (Table 1), martensitic transition temperatures [1, 8, 10, 11] and carried out the magnetic susceptibility measurements [9]. Then, the crystal structure for ZrRh martensite was defined as FeB-type [12]. Later, Semenova and Kudryavtsev [13] detected the complete SME for ZrRh alloy. The structure of martensites of ZrRh and Zrlr alloys was redefined as B19'-type (Table 5) [3, 13]. Koval. Firstov and Kotko have found the nearly complete SME for ZrCu [2]. Alloying and thermal cycling result in the changes of thermal and hysteretic properties and in the increase of shape recovery ratio up to 100% for ZrCu-based alloys [2, 14, 15]. Crystal structure of ZrCu martensites was determined as B19'-type [16, 17] similar to TiNi [18, 19] and ZrRh, ZrIr [13] (Table 1). For  $ZrRu_{(1-x)}Pd_x$  martensites, the authors of [20] claimed that its crystal structure belongs to the CrB-type (Table 5). The description of the crystal structure of the crystallized from amorphous state ZrCu<sub>25</sub>Ni<sub>25</sub> alloy [21] includes a mixture of two different combinations of cubic NiZr and CuZr units (one is like the cubic Cu<sub>2</sub>MnAl Heusler phase (obviously austenitic)

306

**TABLE 1.** Anisotropic temperature factors ( $\beta$ ) and equivalent isotropic factors (*B*) for the *Cm* superstructure. Scale factor  $s = 1.04 \cdot 10^{-5}$ .

Atom	β <sub>11</sub>	β <sub>22</sub>	β <sub>33</sub>	$\beta_{12}$	β <sub>13</sub>	β <sub>23</sub>	$B(\text{\AA}^2)$
Zr	0.09851	0.02423	0.07383	0.00000	-0.04203	0.00000	1.01
Cu	0.04378	0.03796	-0.02738	0.00000	-0.05496	0.00000	0.96

**TABLE 2.** Atomic positions in the unit cell for the *Cm* superstructure. Lattice parameters: a = 0.63374 nm; b = 0.85636 nm; c = 0.53453 nm;  $\beta = 105.58^{\circ}$ .

	position	x	У	z		position	x	У	z
Zr	2a	0.0	0.0	0.0	Cu	2a	0.69309	0.0	0.67810
Zr	2a	0.56362	0.0	0.09805	Cu	2a	0.20635	0.0	0.63303
Zr	4b	0.45808	0.25	0.72281	Cu	2a	0.29125	0.25	0.17479

**TABLE 3.** Anisotropic temperature factors ( $\beta$ ) and equivalent isotropic factors (*B*) for the *Cm* superstructure. Scale factor:  $s = 5.57 \cdot 10^{-4}$ .

Atom	β <sub>11</sub>	β <sub>22</sub>	β <sub>33</sub>	β <sub>12</sub>	β <sub>13</sub>	$\beta_{23}$	Β(Å <sup>2</sup> )
Zr	0.58029	0.02100	0.03022	0.00000	-0.01541	0.00000	0.66
Cu	-0.04882	0.00370	0.02199	0.00000	0.05117	0.00000	0.55

**TABLE 4.** Atomic positions in the unit cell for the  $P2_1/m$  martensite. Lattice parameters: a = 0.32823 nm; b = 0.41484 nm; c = 0.52495 nm;  $\beta = 103.75^{\circ}$ .

	position	x	У	z		position	x	У	Z
Zr	2e	0.38273	0.25	0.20251	Cu	2e	0.11754	0.0	0.63921

and the other is tetragonal). So, there are several definitions of the martensitic crystal structures of Zr-based equiatomic intermetallics. Our aim is to show the important differences of Zr-based equiatomic intermetallics as a novel high temperature shape memory alloys from existing shape memory alloys (SMA), especially TiNi and related compounds.

The Zr-base alloys used in this investigation were arc-melted from iodide zirconium, electrolytic copper, nickel and cobalt in a pregettered argon atmosphere. Good chemical homogeneity was confirmed by X-ray microanalysis (JCXA-733 (Superprobe)). The changes of the hysteresis loops and the characteristic temperatures of the MT were determined by ac electrical resistivity measurements. SME was examined by a bending technique which is described in [22]. Heats of MT were detected with the help of Du Pont 910 and Netzcsh 404 differential scanning calorimeters in the  $-150^{\circ}C \leftrightarrow +600^{\circ}C$  and  $20^{\circ}C \leftrightarrow 1000^{\circ}C$  temperature ranges respectively. Volume changes during MT were determined using Du Pont TMA 934 and THETA DILATRONIC-III dilatometers. Analysis of crystal structure was performed using the X-ray diffractometry (DRON-1.0, Cu $K_{\alpha}$ ) at room temperature and Rietveld refinement (DBWS 9006PC computer program—new version [23]). The transmission electron microscopy (TEM) investigations were performed in a CM20 Twin instrument with a LaB<sub>6</sub> filament and allowing for  $\pm 60^{\circ}$  and  $\pm 30^{\circ}$  tilt ranges around the primary and secondary tilt axes.

# **1.2. Crystal Structure and Martensite Transformation Mechanism**

We are starting with the description of the structure of the ZrCu martensites.

As cast as well as homogenised samples indeed reveal extremely small product plates or variants as shown in Fig. 1a, including the smallest available selected area aperture. Most of the observed SAED patterns in these samples are relatively dense and have rather short first order reciprocal vectors, indicating that they originate from plates with the largest unit cell. Twinned patterns are also often observed. In Fig. 1 some examples of such SAED patterns, obtained from several extensive tilting experiments and combined from different plates, are shown. Keeping the monoclinic symmetry in mind and making the conventional choice for the monoclinic angle between axes *a* and *c*, these diffraction patterns and their relative orientations, shown in the upper right corner of each pattern, yield the following unit cell dimensions for the large structure:

a = 0.64 nm, b = 0.79 nm, c = 0.52 nm, and  $\beta = 105.8^{\circ}$ .

The choice of *c*-axis was taken in reference to the twin planes inside a given plate, the streaking in Figs. 1c and 1d corresponding with the aperiodic nature of these twins. This result is clearly different form the existing suggestion for the large martensite structure [17]. All reflections follow the conditions h+k=2n irrespective of I which corresponds with the monoclinic space groups C2, *Cm* and C2/*m*. The present interpretation allows for the indexing of all reflections and zones as given in Fig. 1. The pattern in Fig. 1c is now [110] zone whereas in the earlier description it was taken as an [010] zone.

As mentioned above, most of the basic SAED patterns reveal short reciprocal vectors. Nevertheless, in some cases some of the low intensity reflections disappear when the sample or selected area aperture is slightly shifted, indicating the existence of a small basic and a large superstructure based on the same lattice. This is shown for the [101], [111] and [112] zones (large cell indexation) in Fig. 2 where the reflections with

308



**Figure 1.** TEM image revealing the small martensite variant (a); different SAED patterns belonging to CM structure and indexed according to the new choice of unit cell (b), (c) and (d); Tilt angles in respect to (b) are also given in the upper right corners.

 $h=2n+_{\Box}1$  and k=2n+1 exist in the patterns of the basic structure (upper row). [100] and [010] zones are the same for both structures. Based on this observation a basic unit cell can be constructed which is again monoclinic but with a  $P2_1/m$  space group and lattice parameters,



**Figure 2.** [201], [221] and [111] zones of the basic cells (italics) (a) and corresponding [101], [111] and [112] of the CM supercell (b). The *hkl* reflections of the latter with h, k=2n+1 are clearly extinct in the former.

a = 0.32 nm, b = 0.39 nm, c = 0.52 nm and  $\beta = 106^{\circ}$ ,

which are similar to those of the small martensite structure as suggested before [17]. Comparing both basic and superstructures it is concluded that the latter consists of two cells of the former in the *a* and *b* directions. The type of distortions occurring during the MT are expected to be the same for both structures although the actual values of the shear angles can differ somewhat.

As the extensive SAED tilt experiments result in alternative unit cell parameters for the large cell, new Rietveld refinement procedure of the old X-ray data, taken at room temperature and including peaks belonging to both product structures, was performed [23, 35, 36]. The lattice parameters and symmetry obtained from the SAED results discussed above were used as starting values for the iteration procedure. After the first step the *R*wp-factor was approximately 35%. Refinement with an isotropic temperature factor yielded *R*wp = 8.8% while changing the latter into an anisotropic one resulted in convergence with *R*wp = 6.3%. The final refinement data as well as the atomic parameters for both product phases are shown in Tables 1 to 4. The corresponding X-ray diffraction profile including the fit



**Figure 3.** X-ray diffraction pattern for ZrCu (points—experiment; calculated curve is from the final Rietveld refinement cycle). A difference curve is also shown.

is shown in Fig. 3.

According to Hill and Howard [36] we estimated the volume fractions for both martensites using scale factors (Tables 2, 4) and obtained 23% for twinned *Cm* and 77% for twinless  $P2_1/m$  martensitic phases. If we take into account the lattice parameter of austenite 0.3262 nm and experimental volume change during forward MT 0.147% measured in [17] we also can calculate the volume change using refined lattice parameters. For the twinned martensite it equal to 0.6% and for the twinless -0.01%. Using the volume fractions we obtain general volume change 0.146% that is very close to experimental value. Reasonability of the proposed new model for the twinned martensite also confirmed by much lower anisotropic temperature factors in comparison with the results obtained in [17].

Using the smaller dimensions of the basic structure, a possible crystallographic relationship with the austenite can now be suggested. Indeed, when comparing the obtained values with those of other martensite structures formed out of *B2* austenite material, an apparent similarity with the monoclinic NiTi *B*19' structure is found [18, 19]. It was suggested before that this structure can be obtained from the cubic *B2* by shearing the (011)*B2* planes, thus forming the monoclinic angle, elongating the [011]*B2* direction and shortening the [100]*B2* and [011]*B2* ones. This is drawn schematically in Fig. 4. The lattice correspondence then becomes

# [100]B2||[100]M, (011)B2||(001)M, (011)B2||(010)M.

Based on the present lattice parameters, no shortening of the [100]*B*2 axis is observed in CuZr. On top of these distortions, some reshuffling of the atoms in NiTi was found after refinement of neutron and X-ray data [7, 8] yielding a  $P2_1/m$  space group.

In addition to these distortions the atoms are shuffled within each unit cell. In Fig. 4 the [010] planes of both martensite structures are shown in comparison to the [011] planes of the parent *B*2 phase. The distortions are clear from the outlines of the unit cells and the relative shifts are indicated by arrows. From this it is seen that no shuffles exist in the  $[011]_{B2}$  direction, *i.e.* every atom remains in its respective  $[011]_{B2}$  or  $(010)_M$  plane. For the supercell the largest displacements are found in the y = 1/4 or 3/4 planes. The doubled lattice parameters for the supercell are primarily caused by the difference in shuffles of the internal Cu atoms on the y = 0 and 1/2 planes.

The results of our further structural analysis are shown in Table 5 with the previously reported parameters of crystal structure for some other *B*2 equiatomic intermetallics given for comparison. The remarkable resemblance in martensitic structures for *B*2 Zr-based equiatomic intermetallics with NiTi and even AuCd (*B*19' and *B*19 respectively) is clearly seen excluding the results reported in [10, 20]. The  $B_{\rm f}$  (CrB) structural type defini-



**Figure 4.** Schematic of the distortions and shuffles bringing the *B*2 austenite into the two martensitic structures.

tion given for  $ZrCo_{36}Ni_{14}$  and  $Zr(Ru_{0.3}Pd_{0.7})$  martensites (Table 5) is unclear because the structural model proposed in [10, 20] depends only on the X-ray powder diffraction [10], limited electron microscopic (EM) studies [20] and possible isomorphism with ZrPt phase without any analysis of the reflections intensities.

In the present work the crystal structure of  $Zr_2CuCo$ ,  $Zr_2CuNi$  and  $Zr_2CoNi$  compounds was refined using the Rietveld procedure [23] similar to the case of ZrCu [24] and a good agreement between experimental and calculated X-ray patterns was obtained (*Rwp* factor was in the range 5–8%). Reasonability of the refined structural parameters for  $Zr_2CuCo$  compound (Table 5) reflects also in a good agreement of the sign and amount of the MT volume effect obtained in dilatometric measurements and calculated using the refined lattice parameters. Unfortunately, the crystal structure of the observed martensitic phase obviously belonging to the first type was not confirmed for  $Zr_2CoNi$  compound because of very small volume fraction (Table 5) and needs additional investigation.

Formation of the two martensitic phases during forward MT significantly differs in *B*2 intermetallics of Zr (excluding Zr<sub>2</sub>CuCo compound) from isomorphous compounds of Ti. The origin of this is will be explained later.

State	Austenitic				∆ <i>v</i> /v, A→M	Volume fraction of martens.	Ref.				
Parameters	<b>a</b> , Å	St. type	<b>a</b> , Å	<b>b</b> , Å	<b>C</b> , Å	β, °	Sp.gr.	St. type	%	%	
TiNi	3.015	B2	2.898	4.108	4.646	97.78	P2 <sub>1</sub> /m	<i>B</i> 19′	-0.02	_	[19]
AuCd	3.322	B2	3.154	4.764	4.864	_	Pmma	<i>B</i> 19	-0.41	_	[28]
ZrCo <sub>36</sub> Ni <sub>14</sub>	3.197	B2	3.253	9.707	4.186	-	Стст	$B_{\rm f}$	+0.28	_	[10]
$Zr(Ru_{0.3}Pd_{0.7})$	?	B2	3.32	10.22	4.48	-	Стст	$B_{\rm f}$	?	_	[20]
ZrRh	3.295	B2	3.31	4.32	5.31	100.4	$P2_1/m$	<i>B</i> 19′	?	100	[13]
Zrlr	3.318	B2	3.33	4.38	5.35	99.9	$P2_1/m$	<i>B</i> 19′	?	100	[13]
ZrCu	3.2624	<b>B</b> 2	1)6.337	8.563	5.345	105.6	Ст		+0.14	23	[16,
			2)3.282	4.148	5.249	103.7	P2₁/m	<i>B</i> 19′		77	17]
Zr₂CuNi	6.45 <sup>@</sup>	L2 <sub>1</sub>	1)6.234	8.973	5.301	103.8	Ст		+0.16	12	[32]
	<sup>@</sup> —[21]		2)3.336	4.193	5.235	103.1	P21/m	<i>B</i> 19′		88	
Zr <sub>2</sub> CuCo	3.2450	<b>B</b> 2	3.275	4.186	5.203	106.2	P2₁/ <i>m</i>	<i>B</i> 19′	+0.20	100	[32]
Zr <sub>2</sub> CoNi	?	?	1)6.439	8.461	5.362	105.8	Ст		+0.07	10	[32]
			2)3.274	4.050	5.340	106.1	P2₁/m	<i>B</i> 19′		90	

**TABLE 5.** Structural parameters of Zr-based and some another intermetallic compounds.

The results of the dynamic mechanical response measurements (DMA) on the second complete MT cycle in ZrCu are presented in Fig. 5. The plate specimen was loaded in parent phase in a way that strain amplitude was equal to about  $10^{-3}$ . It is seen that the low frequency elastic modulus increases linearly on cooling from 300°C until 160°C. Then it starts to decrease and after passing a minimum around 80°C increases again until -50°C. Internal friction (IF) dependence during this cooling shows a broad maximum clearly containing at least two peaks and correlates with a modulus decrease. Further cooling to -50°C reveals a small peak around 0°C. Subsequent heating results in a decrease of the low frequency modulus with a minimum at 310°C with further two step increase that finishes at 370°C. The correspondent IF dependence shows a weak maximum at 50°C and a maximum at 310°C with a prolonged completion.

The elastic modulus and the internal friction on the second complete MT cycle in ZrCu measured by the impulse excitation technique (IET) are shown in Fig. 6. Both properties are showing approximately the same behaviour as for the dynamic mechanical response measurements presented in Fig. 5. High frequency elastic modulus behaviour has almost



**Figure 5.** Evolution of the internal friction (IF) and modulus (*E*) during the second complete MT cycle in ZrCu (5°C/min, DMA—fixed frequency 4 Hz,  $\approx 10^{-3}$  strain amplitude).

the same magnitude and differs only in a slope of its general increase on cooling. IF is significantly lower in this case, especially above 200°C. IF peak on cooling (Fig. 6, closed triangles) correlates well with the DMA one (Fig. 5) but starts higher at 220°C. Then the small peak slightly above 50°C can be seen also. On subsequent heating the high frequency elastic modulus (Fig. 6, open circles) shows a minimum around 280°C that is lower in temperature and smaller in magnitude than the DMA one (Fig. 5). Correspondent IF peak on heating (Fig. 6, open triangles) is very weak. The difference in absolute values in results of DMA and IET measurements is related to the large difference in strain amplitude ( $10^{-7}$  for IET that is  $10^4$  times smaller than for DMA).

The results of the differential scanning calorimetry (DSC) measurements are presented in Fig. 7. The second forward MT exothermic reaction starts at 170°C and finishes around 30°C after passing two maximums. The second reverse MT endothermic reaction starts at 280°C and finishes at 420°C after passing two maximums as well. The general amount of the released heat on the forward MT was equal to  $Q^{A>M} = 977$  J/mol (12.63 J/g) and for the reverse MT it was equal to  $Q^{M>A} = 1345$  J/mol (17.38 J/g). These general two step peaks were fitted with the help of the two distorted Pearson peak functions each and as a result we were



**Figure 6.** Evolution of the internal friction and *E*-modulus during the second complete MT cycle in ZrCu (5°C/min, IET,  $fr \approx 4.2-4.4$  kHz,  $\approx 10^{-7}$  strain amplitude).

able to obtain the transformation heats and temperatures for each step as it is shown in Fig. 7.

Knowing already a presence of two martensitic phases (*Cm* and *B*19') as a result of MT in ZrCu [34] it was obvious to perform in situ high temperature X-ray diffraction measurements in order to bring into correlation structural parameters of all the present phases and there order of appearance with the obtained heats and temperatures of the extracted intrinsic MT steps.

In particular, the Rietveld refinement of X-ray diffraction patterns obtained on cooling and heating during complete MT cycle in ZrCu allowed us to obtain the temperature dependencies of the volume fractions for all phases present as well as changes in there lattice parameters. The results of this analysis are shown in Figs. 8 and 9. It is seen how B2 volume fraction (Fig. 8, open boxes) starts to decrease on cooling at 170°C up to around room temperature correspondingly to the overall exothermic reaction (Fig. 7). At the same time, the volume fractions of the B19' martensite (Fig. 8, closed triangles) and of the martensite belonging to the *Cm* space group (Fig. 8, open circles) increase in amount. It is impossible to recognize which type of martensite forms first but it is clearly seen that B19' martensite finishes its appearance around 100°C, while martensite belonging to the *Cm* space group proceeds to form until about 30°C, replacing austenite completely. It means that the first sharp intrinsic exo-



Figure 7. DSC measurement results of the second complete MT cycle in ZrCu (10°C/min).

thermic peak was caused by the formation of the *B*19' martensite (Fig. 7). Taking into account its volume fraction (23%) and the amount of heat of the first sharp exothermic peak extracted after fitting (343 J/mol) we are able to find the heat of the imaginary separate forward B2 > B19' MT in ZrCu  $Q^{B2>B19'} = 1490$  J/mol.

The same can be done for the prolonged intrinsic exothermic peak that is attributed to the imaginary separate forward  $B2 > C_m$  MT in ZrCu—  $Q^{B2>Cm} = 823$  J/mol. Further heating leads to the appearance of the B2 austenite around 250°C which starts to replace B19' martensite first and corresponds to the sharp peak of the first intrinsic endothermic reaction that finishes around 350°C. The volume fraction of the Cm martensite starts to decrease at 300°C that correlates well with the beginning of the second intrinsic endothermic reaction which is prolonged until approximately 420°C. Again, we are able to obtain MT heats for both reverse in-trinsic MT's. So,  $Q^{B19'>B2} = 2030$  J/mol and  $Q^{Cm>B2} = 1140$  J/mol. According to Ortin and Planes [26], the difference between reverse and forward MT heats represents the amount of energy dissipated during the complete MT cycle  $E_d = Q^{M>A} - Q^{A>M}$ . For the second cycle of MT for ZrCu in general this dissipated energy will be  $E^{\text{ZrCu,d}} = 1345 - 977 = 368$  J/mol. For the imaginary distinct B2 <> B19' and  $B2 <> C_m$  MT's those dissipated energies will be  $E^{B2\leftrightarrow B19',d} = 2030 - 1491 = 539$  J/mol and  $E^{B2\leftrightarrow Cm,d} = 1140$ -823 = 317 J/mol. It is seen that the actual dissipated energy is almost



**Figure 8.** Changes in the volume fractions for *B*2, *Cm* and *B*19' phases derived from the results of the Rietveld refinement of the X-ray diffraction patterns taken on cooling-heating cycle during the second MT in ZrCu.

the same in amount as the possible energy loss for the imaginary distinct  $B2 <> C_m$  MT.

The Rietveld refinement helped us to retrieve the lattice parameters for all the observed phases, also. Temperature dependencies of those lattice parameters are represented in Fig. 9. B2 phase lattice parameter decreases on cooling (and increases on heating) can be simply approximated by the following linear function:  $a_{B2} = 3.2618 + T \cdot 5 \cdot 10^{-5} \text{ Å}$  (Fig. 9a, *T*—temperature, °C). Parameter  $a_{B19}$  increases on cooling (decreases on heating)— $a_{B19'} = 3.2836 - 7.6 \cdot 10^{-5}$  Å. It should be noted that their temperature dependencies cross each other around 200°C that is slightly above  $M_s$  for this complete MT cycle in ZrCu (Fig. 9a). Parameter  $b_{B19'}$  increases almost linearly on heating ( $b_{B19'} = 4.1787 + T \cdot 10^{-4}$  Å), while  $c_{B19'}$ has a maximum at a 350°C as well as monoclinic angle  $\beta_{B19'}$  which decreases on heating (Fig. 9b). This maximum corresponds to  $A_{f1} = A^{B19' \rightarrow B2, f}$  temperature. Lattice parameters  $c_{Cm}$  and  $\beta_{Cm}$  behaviour is very similar to that in B19' martensite, only the peak on heating can be observed around 300°C (Fig. 9b, c). As for the  $a_{Cm}$  and  $b_{Cm}$ , they show the same behaviour on heating (slight increase; Fig. 9c) but on cooling  $b_{Cm}$ undergoes a decrease, while  $a_{Cm}$  increases.

Then, the lattice parameters of the austenite and both martensites were used to calculate the volume per atom changes during B2 <> B19'



**Figure 9.** Lattice parameters changes for *B*2 (c), *Cm* (a, b, c) and *B*19' (b, c) phases derived from the results of the Rietveld refinement of the X-ray diffraction patterns taken on cooling-heating cycle during second MT in ZrCu and MT volume effect calculated from relation between the lattice parameters of the austenite and both martensites (d).

and B2 <> Cm MT's (Fig. 9d). Calculated volume effect for both forward MT's is positive and in the 0.2–0.3% range. Linear approximations of such volume effect temperature dependencies tell us that the B2 > B19' forward MT volume effect is bigger than for B2 > Cm MT in ZrCu. It can be also clearly seen that calculated (Fig. 9d) and observed volume changes during second MT (Fig. 10) are very close to each other.

The results stated above give us the possibility to explain the martensites' formation mechanism in ZrCu. According to the martensites' volume fraction change (Fig. 8) and DSC measurements (Fig. 7) it seems like the *B*19' martensite forms first preferentially. On the other hand, the structural analysis shows that the positive volume effect on forward MT



Figure 10. Dilatometric measurements of the second complete MT cycle in ZrCu.

for B19' is bigger than for *Cm* martensite (Fig. 9d). Taking into account the results of Seo and Schryvers [42] that show alternate  $B19' \rightarrow Cm \rightarrow B19' \rightarrow ...$  'sandwich'-type of microstructure, it is logical to suppose that the formation of B19' martensite at first induces considerable stresses in surrounding matrix and the *Cm* martensite then forms as a relaxation of such stresses. Such forward MT can prevent the plastic deformation and can be the reason for the nearly perfect shape memory effect during MT in ZrCu that is taking place with wide temperature hysteresis [2]. The reverse MT shows that B19' martensite disappears first (Fig. 7, 8). It indicates that MT in ZrCu is non-thermoelastic.

The thermoelasticity concept that was confirmed experimentally [33] includes such requirement that the last portions of martensite phase on forward MT must be transformed to austenite in a first order. In ZrCu we have a completely opposite situation. Such an order of appearance  $\leftrightarrow$  disappearance for the martensites in ZrCu can be also the reason for the more intensive energy dissipation as a result of serious overheating through *B*19' martensite 'barrier' for  $B2\leftrightarrow C_m$  MT.

The most important question in this case, why *B*19' martensite forms first in spite of its bigger MT volume effect, twinless substructure *etc.*? In [24] the lattice correspondence between *B*2 and *B*19' for ZrCu was supposed as  $[100]_{B2}||[100]_{M}$ ,  $(011)_{B2}||(001)_{M}$ ,  $(011)_{B2}||(010)_{M}$ . Lattice parameters  $a_{B19'}$ , and  $a_{B2}$  behaviour (Fig. 9a) shows that their perfect match close to  $M_s$  can favour the formation of *B*19' at first on cooling.

It can be concluded that:

MT in ZrCu is of non-thermoelastic type;

The relaxation of internal elastic stresses during forward B2>B19' MT in ZrCu takes place by means of the formation of additional *Cm* martensite instead of twinning or plastic deformation;

The perfect match of the interplanar spacings of the (100) parallel planes for B2 and B19' phases can be the reason for the initial formation of the B19' martensite.

#### 1.3. Thermodynamic and SME Behaviour

Electrical resistivity and calorimetric measurements show ([2], Table 6) that the hysteresis of MT in ZrCu, Zr<sub>2</sub>CuNi and Zr<sub>2</sub>CoNi compounds is larger in comparison with the case of MT in AuCd, TiNi and even Zr<sub>2</sub>CuCo. The difference in the heats of the reverse and forward MT  $E_d = Q^{M \rightarrow A} - Q^{A \rightarrow M}$  listed in Table 6 (according to Ortin and Planes [26]) represents the energy lost in the complete MT cycle. Comparison of the values of thermal hysteresis and  $E_d$  for MT in B2 Zr intermetallics and AuCd, NiTi and TiPd shows that the irreversible contributions to the non-chemical free energy change in energy balance [26] (friction stresses required to move the interfaces; free energy changes related to defects induced by MT; partial plastic accommodation of the transformational shape and volume changes, mainly accomplished by slip) are higher for Zr-based intermetallics. The temperature intervals of MT ( $M_s-M_f$ ,  $A_s-A_f$ ) that reflect the value of the friction forces are also higher for these compounds (~100 K, Table 6).

The very interesting behaviour of the hysteresis loops was observed on partial cycling during electrical resistivity and confirmed by dilatometric measurements. After stabilisation of the full hysteresis loop (20 thermal cycles) partial reverse MT's result in much lower temperatures of the forward MT on subsequent cooling in  $Zr_2CuNi$  compound (Fig. 5f, the same behaviour was observed for  $Zr_2CoNi$  alloy).

When the reverse partial cycles became more complete the shape of the partial hysteresis loops came nearer to original (full) hysteresis loop. Repeated full cycle exhibited original shape of the hysteresis loop. So, this process is quite reversible in spite of the large width of hysteresis even for the full cycle. Usually, in the case of thermoelastic MT all partial hysteresis loops are internal in respect to the full one (see [27] for example). The same behaviour can be seen for  $Zr_2CuCo$  compound (Fig. 13a).

SME for Zr<sub>2</sub>CuNi (the same as the SME behaviour in Zr<sub>2</sub>CoNi) and Zr<sub>2</sub>CuCo compounds are shown in Fig. 11 and Fig. 12 respectively. Shape recovery ratio ( $K_{SME}$ ) for Zr<sub>2</sub>CuNi is nearly complete (~ 90%) similar to ZrCu ([2], Table 6) contrary to wide temperature hysteresis, large amount of  $E_d$  and high temperatures of shape recovery. Decrease of the hysteresis size and  $E_d$  reflect in complete SME for Zr<sub>2</sub>CuCo compound (Table 6, Fig. 12).

High reversibility of MT in Zr-based intermetallics was confirmed by the stabilization of the hysteresis loops on thermal cycling and perfect SME. This reversibility was observed simultaneously with the high level of the irreversible processes during MT in this compounds. This unusual situation

Compound	M₅, °C	M <sub>f</sub> , °C	A <sub>s</sub> , ∘C	A <sub>f</sub> , °C	<i>T</i> ₀, °C	∆ <i>T</i> , °C	Q <sup>A</sup> → <sup>M</sup> , J/g	Q <sup>M→A</sup> , J/g	<i>E</i> <sub>d</sub> , J/g	K <sub>sme</sub> , %	Ref.
AuCd	60	55	75	80	70	20	1.87	2.22	0.35	100	[29]
TiNi	50	20	60	90	55	40	27.3	29.9	2.6	100	[30]
TiPd	560	550	580	590	570	40	33.2	33.7	0.5	100	[31]
Zr <sub>2</sub> CuCo	-25	-125	-65	25	-42	55	14.2	15.8	1.6	100	this work
ZrCu	140	40	230	330	185	175	13.9	19.1	5.2	~ 90	[2], this work
ZrRh	450	350	550	650	500	200	_	_	_	100	[13]
Zr₂CuNi	610	510	690	790	650	180	35.2	40.1	5.3	~ 90	this work
Zr <sub>2</sub> CoNi	620	480	700	950	710	225	29.3	35.7	6.4	~ 90	this work
Zrlr	750	700	890	990	830	215	-	-	_	_	[13]

**TABLE 6.** Thermodynamical parameters of MT and shape recovery ratio for Zrbased and some another intermetallics.

may be explained as follows. High values of the friction forces in Zr<sub>2</sub>CuCo alloy (wide temperature intervals of MT, Table 6) may be attributed to the influence of the even small but positive volume effect of the forward MT. In the cases of MT in ZrCu, Zr<sub>2</sub>CuNi and Zr<sub>2</sub>CoNi compounds additional irreversible contribution to the non-chemical free energy change in energy balance becomes important — interaction of the martensitic crystals belonging to the different kinds of martensites that can be also the source of the unrecoverable plastic deformation resulted in nearly complete SME ( $K_{SME} \cong 90\%$ , Table 6). It was indirectly confirmed by the unusual behaviour of partial hysteresis loops in the cases of Zr<sub>2</sub>CuNi and Zr<sub>2</sub>CoNi compounds when the two martensitic phases were formed (Table 5, Fig. 13f). On partial MT cycles in Zr<sub>2</sub>CuCo alloy the situation is very close to the thermoelastic case (Fig. 13a) due to the formation of only one kind of martensite (Table 5).

As for the case of ZrRh alloy having complete SME and wide hysteresis [13] when the presence of only one martensite was reported it should be mentioned that we can find the similar situation in [10]. The authors of this paper also reported about the one martensite with the results of dilatometric measurements with two stages of reverse MT. It can be supposed that the more detailed structural analysis will show the presence of two martensites in these alloys.

Drastic change of MT behaviour from thermoelastic in Zr<sub>2</sub>CuCo to non-



Figure 11. SME in Zr<sub>2</sub>CuNi compound.

thermoelastic in  $Zr_2CuNi$  initiated our interest to the changes along the  $Zr_2CuNi$ – $Zr_2CuCo$  quasi-binary cross-section.

The results of electrical resistivity measurements are shown in Figs. 13, 14, 17. It is seen that the MT in the  $Zr_2CuNi$  compound takes place



Figure 12. SME in Zr<sub>2</sub>CuCo compound.

with a wide hysteresis (180 K) and MT intervals (100 K). Co for Ni substitution results in a linear decrease of MT temperatures by  $\approx 600$  K (Fig. 14) and general hysteresis decrease to 50 K for Zr<sub>2</sub>CuCo with a maximum at 15–20 at.% of Ni (Fig. 17). The  $A_s$  temperature, higher than  $M_s$  at high Ni content, becomes lower than  $M_s$  on Co additions (Fig. 14). After stabilization of the full hysteresis loop (20 thermal cycles) partial reverse MT's in  $Zr_2CuNi$  result in much lower  $M_s$ ,  $M_f$  temperatures on subsequent cooling as it was mentioned above (Fig. 13f). When the reverse partial cycles became more complete the shape of the partial hysteresis loops came nearer to the original (complete) hysteresis loop. A new full cycle showed again the original shape of the hysteresis loop. Addition of 4.42 at.% of Co weakens the effect mentioned above with the simultaneous appearance of the effect of the reverse MT temperatures increase on partial cycling (Fig. 13e, thin dashed line-1st partial cycle, thin solid lines—subsequent cycles). It is clearly seen from Fig. 13d that partial cycling results in disappearing of the forward MT lowering effect for the alloy with 9.6 at.% Co. The starting partial reverse path (Fig 13d, thin dashed line) shows usual behaviour but the subsequent partial reverse paths are outside of the stabilized complete hysteresis loop. The same behaviour was observed for the alloys containing 14.8 (Fig. 13c) and 19.9 (Fig. 13b) at.% of Co. It can also be seen (Fig. 13b, c, d, e) that every subsequent partial path is outside of (higher in temperature than) the previous one. Moreover, after the finishing of partial cycling the subsequent first complete reverse path is much higher in temperature even compared to the previous partial paths (Fig. 13b, c, d, bold dashed line). A final complete cycle has the same shape as the original stabilized one (Fig. 13b, c, d, e) similar to Zr<sub>2</sub>CuNi (Fig. 13f).

X-ray diffraction measurements confirm the presence of only one B19' type martensite in  $Zr_2CuCo$  and simultaneous formation of two martensitic phases in  $Zr_2CuNi$  belonging to the  $P2_1/m$  (B19' type) and Cm space groups similar to ZrCu [24]. The lattice parameters are not changed significantly along the  $Zr_2CuCo-Zr_2CuNi$  cross-section because the atomic radii of Co and Ni are nearly identical. Considerable changes can be seen for *Cm* and  $P2_1/m$  martensite volume fractions (Fig. 15) that were obtained according to [6]. The amount of the B19' type martensite volume fraction increases from  $\approx 15\%$  in  $Zr_2CuNi$  to  $\approx 95\%$  in the alloy with 9.6 at.% of Co. The exact structural parameters for the studied cross-section will be published elsewhere.

SME for the Zr<sub>2</sub>CuNi–Zr<sub>2</sub>CuCo quasi-binary cross-section was studied under the 120 MPa static load applied before cooling below the  $M_s$  temperature. Then the specimens were released at 77 K after storing of 0.2–0.5% martensitic deformation for subsequent free shape recovery on heating. The composition dependence of the shape recovery ratio  $K_{\text{SME}} = (\epsilon_M/\epsilon_R) \cdot 100\%$  is shown in Fig. 16. It can be seen that the shape recovery



**Figure 13.** Evolution of the MT hysteresis loops during full and partial thermal cycling along the  $Zr_2CuCo-Zr_2CuNi$  quasibinary cross-section (Ni for Co substitution).



**Figure 14.** MT temperatures as a function of Ni content for the  $Zr_2CuNi$ – $Zr_2CuCo$  quasibinary cross-section.



Figure 16. Shape recovery ratio vs Ni for Co substitution for the  $Zr_2CuCo-Zr_2CuNi$  quasibinary cross-section.



**Figure 18.** Martensite transformation heats as a function of Ni content for the  $Zr_2CuNi-Zr_2CuCo$  cross-section.



**Figure 15.** *Cm* and  $P2_1/m$  martensite volume fractions changes vs Ni for Co substitution.



**Figure 17.** MT hysteresis changes vs Ni for Co substitution for the  $Zr_2CuCo-Zr_2CuNi$  cross-section.



**Figure 19.** The energy dissipated on MT complete cycle  $(E_d = Q^{M \rightarrow A} - Q^{A \rightarrow M})$  vs Ni for Co substitution.

becomes complete on Co additions through a minimum at 15–20 at.% of Ni ( $K_{\text{SME}} \ge 85\%$ ).

The results of the compression tests with subsequent SME testing in a dilatometer are shown in Figs. 20 and 21 respectively. It is seen that with Ni additions the strengthening during deformation increases that resulted in a cracking of alloy with 20.31 at.% of Ni content when the deformation exceeds 5% (Fig. 20). Shape recovery for most alloys studied is not smaller than 1% excepting  $Zr_2CuNi$  (Fig. 21).

The results of calorimetric measurements are shown in Figs. 18, 19, 22, 23. Considerable values of MT heats for the alloys with high Ni content (Fig. 18) are noticed. Co additions significantly decrease the transformation heats by  $\approx$  2 kJ/mol. The difference in the reverse and forward MT heats that represents the energy lost in the complete MT cycle  $E_d = Q^{M \rightarrow A} - Q^{A \rightarrow M}$  (according to [26]) is shown in Fig. 19.  $E_d$  values also decrease with increasing Co content through the maximum at  $\approx 20$  at.% of Ni and correlate well with the thermal MT hysteresis behaviour (Fig. 17) and the minimum in shape recovery (Fig. 16). Exo- and endothermic reaction peaks during MT along Zr<sub>2</sub>CuCo-Zr<sub>2</sub>CuNi quasi-binary crosssection are shown in Fig. 22 and 23 respectively. Prolonged forward MT intervals for the alloys with higher Ni content are clearly divided into two overlapped stages (Fig. 22) as well as the reverse MT's (Fig. 23). The difference exists only in the actual temperature positions of the major and minor peaks. In Zr<sub>2</sub>CuNi and alloys with 20.31 and 15.51 at.% Ni, the second minor stage of the forward MT is lower in temperature than for the major stage (Fig. 22) in contrast with the reverse MT when the temperature position of the minor peak is higher (Fig. 23). The situation is changed drastically for the case of MT in the alloy containing 10.22 at.% Ni. The minor stage of the reverse MT appears at temperatures lower than the major stage (Fig. 23) similar to the forward MT (Fig. 22). Further Co additions result in one stage behaviour for forward and reverse MT's.

Such correspondence between the different stages of forward and reverse MT's gives an opportunity to conclude that the MT in alloys with high Ni content is non-thermoelastic. The thermoelasticity concept that was confirmed experimentally [33] includes such requirement that the last portions of martensite phase on forward MT must be transformed to austenite in a first order. The minor peaks observed due to the formation of two different interacting martensite phases are lower in temperature than the major peaks (Fig. 22) and consequently they must be also lower than the major peaks on the reverse MT in the thermoelastic case. In reality, we have the opposite situation as was mentioned above (Fig. 23). Thus, a very high level of dissipative processes in a latter case can be influenced only by the elastic interaction between martensitic crystals of different type. Internal stresses produced by the major kind of martensitic crystals suppress the reverse MT for another kind of martensite.



**Figure 20.** Compression tests for alloys along the  $Zr_2CuCo-Zr_2CuNi$  quasibinary cross-section (Ni for Co substitution) before the SME testing in the dilatometer.



Figure 21. SME testing of the alloys along the  $Zr_2CuCo-Zr_2CuNi$  quasibinary cross-section (Ni for Co substitution) in a dilatometer.



**Figure 22.** Evolution of exothermic reaction peaks during forward MT along the  $Zr_2CuCo-Zr_2CuNi$  quasi-binary cross-section (Ni for Co substitution); arrows indicate the position of the second stage of the forward MT.

This explanation is valid for the case of *Cm* phase majority in  $Zr_2CuNi$  where the major forward MT is prolonged as for an alloy with 15.51 at.% Ni with the major *B*19' type phase having the sharp forward MT exothermic peak (Fig. 22). Only at 14.8 at.% Co, the situation becomes very close to the thermoelastic case. This can be concluded based on the observed sequence of the major and minor MT stages (last portions of martensite (minor kind) are transformed first into austenite).

The value of the internal shear stresses that can decrease the partial forward MT by 100 K can be estimated (Fig. 13f, upper partial path). The Clausius–Clapeyron like equation can be rewritten as follows

$$d\tau = dT \cdot dQ / (T_0 \cdot V_{\text{mol}} \cdot \gamma),$$

where  $d\tau$  is the internal shear stress change associated with the martensite crystals, dT = 100 K is the partial forward MT temperature change,  $dQ = MQ^{A \rightarrow M} = 0.2 \times 2320$  J/mol = 464 J/mol (*M* is the volume fraction of the martensite that will be formed on partial forward MT;  $Q^{A \rightarrow M}$  transformation heat of the complete forward MT),  $T_0 \approx 925$  K is the phase equilibrium temperature,  $V_{mol} = 10.25 \cdot 10^{-6}$  m<sup>3</sup>/mol is the molar volume of Zr<sub>2</sub>CuNi and  $\gamma = 0.1a/b = 0.01$  is the shear strain (*a/b*—martensitecrystals aspect ratio) and, finally,  $d\tau = 0.489$  GPa. Then, the shear stresses that can be produced by martensitic crystal in surrounding aus-



**Figure 23.** Evolution of endothermic reaction peaks during reverse MT along the  $Zr_2CuCo-Zr_2CuNi$  quasi-binary cross-section (Ni for Co substitution); arrows indicate the position of the reverse MT stage that can be put into the correspondence with the second stage of the forward MT.

tenite can be estimated from the following expression:  $\tau = G\gamma = 0.5E\gamma$ . where G and  $E \approx 100$  GPa [34] are the shear and Young's moduli and we achieve  $\tau \approx 0.5$  GPa. So, it means that after partial reverse cycle small austenitic 'islands' are imposed by elastic stresses produced by the neighbouring martensitic crystals and their influence is enough to suppress the subsequent forward partial MT in a way that we have observed (Fig. 13f). Naturally, the martensite crystals produce short range stresses in a matrix. That is why the maximal effect of partial forward MT decrease was achieved at smaller degrees of partial reverse MT. The shape of partial hysteresis loop came nearer to the shape of complete cycle at higher degrees of partial reverse MT because the austenite becomes more free from the influence of internal stresses. Thus, it can be concluded that the unusual behaviour on partial thermal cycling in Zr<sub>2</sub>CuNi compound may be attributed to the elastic interaction between martensitic crystals of different type and austenite due to the observed effect of the restoration of the original shape of the complete hysteresis loop after partial cycles.

Disappearance of the effect described above on Co additions is combined with the reverse partial paths of temperature increase that can be attributed to the significant changes in martensites volume fractions (Fig. 15). It should be fairly noted that all the estimations can be applied even for the thermoelastic MT in  $Zr_2CuCo$  or in another known SMA where such effects are absent or very weak. It means that one has to take into account not only the volume fractions of two different martensites but also their configurations in respect to each other. This task needs quantitative metallographic measurements that are now in progress.

# 1.4. Influence of Alloying and Thermal Cycling on the Martensitic Transformation and Shape Memory Effect in ZrCu Intermetallic Compound

Figures 24a, 26a and 28a show the results of the electrical resistivity *vs* temperature measurements during alloying by Ni, Ti and NiTi respectively. Additions of the Ni lead to the general increase of the Ms temperature up to 510°C (19.85 at.% of Ni) with a small minimum at 2.1 at.% of Ni (Fig. 24a, solid line). Titanium and NiTi additions, on the other hand, decrease the Ms (Figs. 26a, 28a, solid line). The hysteresis of the MT in all cases passes through a maximum (Figs. 24a, 26a, 28a, dashed line).

At higher concentrations of Ni hysteresis decreases to the approximately initial value, ~  $200^{\circ}$ C, whereas for Ti and NiTi additions, the hysteresis drops to even lower values, ~  $50^{\circ}$ C (at 12.91 and 39.81 at.%, respectively).

6500



а HV 30, MPa 6000 5500 5000 4500 4000 1.0Deformation, % b 0.5  $0.0 \stackrel{
m L}{0}$ 1 10 20 Ni content, at.%

**Figure 24.** Hysteresis,  $M_s$  (a) and shape recovery ratio (b) behaviour *vs* Ni content.

**Figure 25.** Vickers hardness (a) and martensitic deformation stored under 100 MPa (b) behaviour *vs* Ni content.





**Figure 26.** Hysteresis,  $M_s$  (a), and shape recovery ratio (b) behaviour *vs* Ti content.

**Figure 27.** Vickers hardness (a) and martensitic deformation stored under 100 MPa (b) behaviour *vs* Ti content.

SME examination shows the increase of the shape recovery to 100% in the case of alloying by Ti and NiTi (Figs. 26b, 28b). The influence of Ni is shown in Fig. 24b. It is seen that the complete SME was observed only at ~ 2–5 at.% of Ni when the curve passes through its maximum. Further alloying by Ni leads to the decrease of shape recovery to ~ 40% at 19.85 at.% of Ni. Martensitic deformation stored under the 100 MPa load during forward MT for all alloying cases is shown in Figs. 25b, 27b and 29b. The martensitic deformation, from this figures, is seen to drop from a value of 0.9% for ZrCu to 0.1–0.3% when nickel and titanium are added to this alloy.

The results of the Vickers hardness measurements are shown in Figs. 25a, 27a and 29a. The *HV*30 of the martensite in ZrCu is nearly 5000 MPa. Alloying with nickel reduces the *HV*30 to about 4500 MPa, through a maximum of 6000 MPa is observed for an alloy with 2.1 at.% of Ni. Titanium and NiTi additions, on the other hand, harden the ZrCu. For NiTi additions the *HV*30 value passes through a maximum (~ 7000 MPa) at 21.7 at.% of NiTi but decreasing subsequently with further addition to the initial unalloyed value (Fig. 29a, solid line). *HV*30 of austenite at ~ 30 at.% of NiTi addition is smaller than that of martensite but some increase (not more than 1000 MPa) with further addition is observed (Fig. 29a, dashed line). In the initial stages, changes in the *HV*30 of martensite with titanium additions is very similar to that of NiTi additions (Fig. 27a, solid





**Figure 28.** Hysteresis,  $M_s$  (a), and shape recovery ratio (b) behaviour *vs* NiTi content.

**Figure 29.** Vickers hardness (a) and martensitic deformation stored under 100 MPa (b) behaviour *vs* NiTi content.

line). At later stages the behaviour becomes more complex.

The changes in the stored martensitic deformation and *HV*30 correlate well with the changes in *HV*30 produced on alloying. An increase in hardness corresponds to a decrease in the martensitic deformation (Figs. 25, 27, 29). Since the hardness is proportional to the yield stress, it is most likely that the increased yield stress affects the martensitic deformability.

As for the shape recovery decrease in the case of alloying with Ni, it may be attributed to the negative slope of the yield stress temperature dependence, because we was obliged to load our specimens above  $M_s$ , which rises with the increase of the Ni content.

According to the views developed in [43] the martensitic transition temperatures of the TiMe (Me: Metal) compounds depend on the ratios of the lattice parameters and the density of states (N(E)) at the Fermi level in TiMe and in pure Ti. The increase of the N(E) at the Fermi level destabilizes *B2* phase in respect to martensite and vice versa. It has been shown that the similar dependence is valid for ZrMe compounds as well. Thus, the changes in the electronic concentration (e/a) must lead to changes in the values of N(E) at the Fermi level resulting in the  $M_s$  shift. So, the  $M_s$  increases on alloying with Ni (Fig. 30, solid line), which has an atomic radius close to Cu and a higher e/a, implying that Ni additions

400

300

200

100



Temperature, °C -100 -200 0 5 10 Ti content, at.%

Figure 30. Characteristic temperatures of the MT vs Ni content.

Figure 31. Characteristic temperatures of the MT vs Ti content.

М.

 $-M_{\rm f}$ 

 $A_{i}$ 

15

must result in the increase of the N(E) at the Fermi level.

Alloying with Ti and NiTi leads to an  $M_s$  drop (more intensive in the Ti case; see Figs. 26a, 28a, solid line). In contrary to Ni, Ti has an identical e/a and smaller atomic radius. It has been shown [44] that the alloying with an element having the same e/a, is equivalent to the influence of external pressure. Describing the effect of pressure on the thermal hysteresis of



Figure 32. Transformation heats of the MT vs Ni content.



Figure 33. Transformation heats of the MT vs Ti content.



**Figure 34.** Comparison of the hysteresis of the MT and energy dissipated on the complete cycle *vs* Ni content.

**Figure 35.** Comparison of the hysteresis of the MT and energy dissipated on the complete cycle *vs* Ti content.

a first-order transition, Das and Ghosh [45] have shown that the thermal hysteresis becomes progressively larger with increasing pressure. It may then be supposed that the substitution of Zr by Ti leads to a widening of the thermal hysteresis of the MT in ZrCu with alloying additions. On the other hand, Wang and Ernst [5] have shown that the stability of the B2 phase in respect to martensite rises in ZrMe $\rightarrow$ TiMe equiatomic compounds row. This fact would also explain the decrease in the martensitic temperatures as Ti is substituted for Zr in ZrCu.

In order to check the validity of our previous assumptions, a systematic study was carried out by means of paramagnetic susceptibility measurements, X-ray diffractometry, and X-ray photoelectron spectroscopy (XPS) in order to determine crystal and electronic structure changes along with alloying additions. The experimental MT temperatures were compared with those calculated for ZrCu-based alloys with the help of Shabalovskaya semiempirical approach [43]. The results of such a comparison are discussed below.

According to [45] the ratio between the lattice parameters of the high temperature phase of the concerned TiMe compound and pure Ti is the term of an equation for  $M_s$  which defines TiMe B2 phase stability. In our case we shall consider the ratio between lattice parameters of the B2 phase of ZrCu-based alloys and the lattice parameter of the  $\beta$ -phase of pure Zr. Experimental results are shown in Fig. 36. It can be seen that alloying with Ni replacing Cu does not result in any significant changes in lattice parameter of the B2 phase at 1173 K. On the contrary, Ti replacing Zr additions result in a linear decrease ( $a_{B2}$ =3.260<sub>3</sub>-3.2·10<sup>-3</sup>C<sub>Ti</sub> Å, C<sub>Ti</sub>-



**Figure 36.** *B*2-lattice parameters and  $a_{ZrMe}/a_{Zr}$  ratio *vs* alloying of ZrCu compound.



**Figure 37.** Paramagnetic susceptibility and  $\chi_{ZrMe}/\chi_{Zr}$  ratio *vs* alloying of ZrCu compound.

at.% Ti) of the room temperature *B*2-phase lattice parameter. In the case of alloying by Ti we used room temperature lattice parameter of Zr  $\beta$ phase [47] ( $a_{\beta}^{RT} = 3.5658_5$  Å) for  $a_{(Zr,Ti)Cu}/a_{Zr}$  calculations. In the case of alloying by Ni we measured lattice parameter of Zr  $\beta$ -phase at 1173 K ( $a_{\beta}^{1173 \text{ K}} = 3.6835_3$  Å) for  $a_{Zr(Cu,Ni)}/a_{Zr}$  calculations. The ratio itself is about 0.9 in magnitude (Fig. 36) and remained almost unchanged. Nevertheless, in both cases of alloying, a linear dependence of the austenite lattice parameters takes place. It means that, contrary to Shabalovskaya findings for TiMe compounds (parabolic dependence of the austenite lattice parameter on  $M_s$ ) [43], we have to make changes for the ZrMe  $M_s$  expression as follows.

# $M_{\rm s} = M_{\rm s}(Zr)(a_{\rm ZrMe,B2}/a_{\rm Zr,b.c.c.})(\Delta E/W_{\rm Me})[N_{\rm d}(E_{\rm F})_{\rm ZrMe}/N_{\rm d}(E_{\rm F})_{\rm Zr}],$

where  $M_{\rm s}({\rm Zr}) = 1136$  K, the temperature of the  $\beta > \alpha$  MT in pure Zr; *a* is the lattice parameter of the cubic *B*2 or  $\beta$ -phase;  $\Delta E$  is the shift of the maximum of the electronic density of states (*e*-DOS) of Me (in our case Cu and Ni) in the compound with respect to that in the elemental state;  $W_{\rm Me}$  is the Me *e*-band width in ZrMe compound;  $N_d(E_{\rm F})$ 's are the Zr *d*-DOS's at Fermi level  $E_{\rm F}$ .

Since ZrCu is paramagnetic [11], its magnetic susceptibility is directly proportional to Zr *d*-DOS at Fermi level according to Stoner expression [48]. It means that we can use the ratio of the ZrMe and pure Zr magnetic susceptibilities instead of DOS in the equation above. The results of room temperature magnetic susceptibility measurements ( $\chi$ ) are shown in Fig. 37. It can be clearly seen that alloying by Ti does not change  $\chi$  too much, while Ni additions significantly increase the paramagnetic susceptibility of ZrCu. Taking into account room temperature magnetic susceptibility of pure Zr (120.10<sup>-6</sup> cm<sup>3</sup>/mol) we found the ratio  $\chi_{ZrMe}/\chi_{Zr}$  for Ti additions to be



**Figure 38.** XPS valence band of the ZrCu compared with pure Cu and Zr;  $\Delta E$  shift is also shown.

**Figure 39.** XPS valence band of the ZrCu along with Ni instead of Cu additions;  $\Delta E$  shifts for Zr–Cu–25.1 at.% Ni in respect with pure Cu and Ni spectra are also shown.

about  $\sim_{\Box}$  0.5 in magnitude, while Ni additions result in the increase of such ratio from  $\sim$  0.5 up to  $\sim$  2.2.

 $\Delta E$  and  $W_{Me}$  were obtained using XPS measurements of the valence band of ZrCu quasi-binary intermetallic compounds. The results are shown in Figs. 38–41. It can be seen for the binary ZrCu compound (Figs. 38, 40, 41) that the electronic band shift in respect to pure Cu is about 0.93 eV, while the band width is about 1.6 eV. Alloying by Ti does not affect XPS valence band spectra in a very significant way though some changes were detected as a result of a peak fitting procedure (Fig.



**Figure 40.** Valence band parameters of the ZrCu along with Ti replacing Zr additions.

**Figure 41.** Valence band parameters of the ZrCu along with Ni replacing Cu additions.



**Figure 42.** Experimental [4] and calculated  $M_{\rm s}$  behaviour along with Ti replacing Zr additions.

**Figure 43.** Experimental [4] and calculated  $M_{\rm s}$  behaviour along with Ni replacing Cu additions.

40). Ni additions result in more visible alterations of the valence band (Fig. 39). In addition to the main peak in ZrCu, located around –3.5 eV, another peak arises closer to Fermi level (around –1.9 eV) due to Ni additions. So, in this case we considered a specific value of  $\Delta E$  taken as a weighed shift of the two peaks mentioned above in respect to pure Cu and pure Ni XPS spectra (Fig. 39). The  $W_{Me}$  was assumed to increase linearly from 1.6 eV in ZrCu to 3.25 eV in Zr–(Cu–25.1 at.% Ni) (width value was taken for both peaks together (Fig. 39)). As a result, we found that for both cases of Ti and Ni alloying the  $\Delta E/W_{Me}$  ratio behaves in a similar way decreasing from ~ 0.57 to ~ 0.45 (Figs. 40, 41).

Finally, we were able to calculate Ms values using the formula presented above for the binary ZrCu and ZrCu-based quasi-binary intermetallic compounds containing Ti and Ni. The results of such calculations compared with  $M_{\rm s}$  experimental data [14] are shown in Figs. 42 and 43 respectively. It can be seen that the absolute calculated  $M_{\rm s}$  values dovetail into experimental data well. In this sense, all terms of the expression for  $M_{\rm s}$  of ZrMe quasi-binary compounds are of the same importance. On the other hand, in the case of Ni alloying only the term, that corresponds to an increase of Zr d-states at Fermi level ( $\chi_{ZrMe}/\chi_{Zr}$ ) due to their redistribution from the depth of the valence band, governs  $M_{\rm s}$  increase or, in other words, instability of the high-temperature B2 phase. In the case of the binary ZrCu and Ti alloying, the  $\Delta E/W_{Me}$  term, which is controlling interatomic interaction in compounds, determines the  $M_{\rm s}$  decrease or, in other words, B2-phase stability. In this case bonding increases due to the enhancement of sp-d hybridization. The quantum mixing of the orbitals results in suppression of the tendency towards shear transformation as well as it was pointed out in [45] for TiMe compounds.

In all of alloying cases the increase in shape recovery is accompanied



**Figure 44.** Structure of the ZrCu compound after the 90 cycles in  $-196-400^{\circ}$ C temperature range and subsequent heating to  $400^{\circ}$ C (*B*2 austenite): dark field in (100) type reflection (a); diffraction pattern, zone axis [100] (b); diffraction pattern, zone axis [110] (c); diffraction pattern, zone axis [113] (d).

by a widening of the hysteresis loop, a situation similar to that occurring upon thermal cycling of the equiatomic ZrCu [2]. Some structural changes are responsible for such situation in the latter case. So, let us describe these changes in comparison with MT and SME behaviour during thermal cycling of ZrCu.

The results of electron microscopic investigation at room temperature of the ZrCu specimens in thermally cycled state (after 90 cycles in  $-196 - 400^{\circ}$ C temperature range with the heating-cooling rate  $50 - 100^{\circ}$ C/sec) are shown in Figs. 44, 45.

In the austenitic state (after the heating to  $400^{\circ}$ C, Fig. 44) the formation of the cellular substructure with regular dislocations ensembles of high density were revealed. The dislocations (Fig. 44a) are oriented perpendicular to the diffusive streaks running to (100) type reflections (Fig. 44b) and (110) type reflections (Fig. 44c). Appearance of the weak spots of  $\frac{1}{2}(100)$  and  $\frac{1}{2}(110)$  type (Fig. 44b) can be attributed to the formation of zones with lower translation symmetry than in the major *B*2 structure.

In the martensitic state (after the cooling to  $-196^{\circ}$ C) only the thin plates (10–20 nm thickness) of the *B*19' martensite (Fig. 45a) can be found. These thin plates are in the twinning correspondence ((011)<sub>*B*19'</sub> twinning plane) to each other similar to the as-cast condition [2] and filling



**Figure 45.** Structure of the ZrCu compound after the 90 cycles in  $-196-400^{\circ}$ C temperature range and subsequent cooling to  $-196^{\circ}$ C (*B*19' martensite): bright field from the orthogonal oriented ensembles of martensite crystals (a) and corresponded diffraction pattern, zone axis [100] (b); bright field from the neighbouring martensite and retained austenite (c) and corresponded diffraction pattern, zone axis [100] (d).

the cells seen in the austenitic state (Fig. 44a). So, it can be also seen that the thickness of the B19' martensite plates undergoes ten fold decrease during thermal cycling in comparison with the as-cast condition [2]. Figures 45c and 45d represent the neighbouring austenite and martensite regions and corresponded diffraction pattern. Analysis of this pattern gives us the possibility to identify the orientation relationship between austenite and B19' martensite for ZrCu

 $[100]_{B2}||[100]_{B19'}, (010)_{B2}||(011)_{B19'},$ 

that is in good agreement with the scheme of the distortions and shuffles for  $B2 \rightarrow B19'$  structure change given in section 1.2 of this article (Fig. 4, [24]).

We have also studied the MT and SME behaviour in equiatomic ZrCu on thermal cycling by ac electrical resistivity method and bending technique [22] correspondingly. The results are shown in Figs. 46–48. Evolu-



**Figure 46.** The changes of the MT hysteresis loop in ZrCu during thermal cycling: 1) 2nd cycle, 2) 45th cycle, 3) 90th cycle.

tion of the MT hysteresis loops shows (Fig. 46) that at the initial stage of thermal cycling (up to 50 cycles) the widening of the hysteresis loop takes place (from 180°C to 300°C) without significant changes in temperature intervals of MT (~100°C) and in the temperature position of the geometric centre of the full hysteresis loop (or  $T_0$  position). Subsequent thermal cycling results in serious lowering of the  $M_s$  temperature with continuous widening of the hysteresis that takes place close to the -10°C (Fig 46, 90th cycle, insert).

The similar behaviour was observed (Fig. 47) during DSC measurements (performed at 10°C/min heating-cooling rate). The splitting of the MT into two stages with the increase of the number of thermal cycles is clearly seen (Fig. 47a–e). The difference between the DSC and electrical resistivity measurements results in the number of cycles necessary for the splitting mentioned above caused by the different heating-cooling rates used. Nevertheless, the transformation temperatures of the MT with narrow hysteresis correlate well for both methods. The heats of the transformation for this transition are very close in values to the thermoelastic MT's (Table 6, Fig. 47 f, g). So, it can be supposed that the shape recovery should be complete for this transition.

The results of SME measurements are shown in Figs. 48, 49. Samples in as-cast state demonstrate nearly complete SME (85–90% of shape recovery ratio) (Fig. 48). The amount of martensitic deformation was equal to 0.5–1.3%, which was stored under the external stress 50–150 MPa. It was found that the recovery stress generation takes place during reverse MT on heating (Fig. 48, dashed line). Shape recovery ratio in this



**Figure 47.** Evolution of DSC curves for ZrCu in  $-100^{\circ}C-400^{\circ}C$  temperature range: a) 2nd cycle cooling, b) 4th cycle cooling, c) 6th cycle cooling, d) 8th cycle cooling, e) 10th cycle cooling, f) 12th cycle cooling, g) 12th cycle heating.

case is lowered to 60%.

SME behaviour of thermally cycled specimens is shown in Fig. 49. It is seen that both of the MT stages are accompanied by complete SME excluding the situation when the sample restores its shape in the presence of external load (10% of shape recovery ratio only). Complete SME in the case of MT with narrow hysteresis is realized simultaneously with the generation of considerable recovery stresses (up to 300 MPa).

It should be remarked that the stored martensitic deformation under 150 MPa in specimens after thermal cycling treatment is smaller than in as-cast state. MT with narrow hysteresis exhibits 0.1% and 0.4% of deformation is accumulated in the wide hysteresis case.

So, the influence of thermal cycling is expressed by an increase of shape recovery ratio (without external load) and widening of the hysteresis loop. Miyazaki, Igo, and Otsuka [50] explaining the effect of the thermal cycling in various shape memory alloys considered the following



**Figure 48.** SME behaviour at the second thermal cycle (solid line—sample was unloaded before reverse MT, dashed line—shape recovery under external load), P = 150 MPa.

causes: a) change in the degree of order, b) ageing effect during thermal cycling, c) introduction of dislocations.

All of the causes mentioned may result in widening of the MT hysteresis loop.

On the other hand, the precipitation processes can harden the matrix and



**Figure 49.** SME behaviour at the 90th thermal cycle (solid line—sample was unloaded before reverse MT, dashed line—shape recovery under external load), P = 150 MPa.

prevent it from the irreversible plastic deformation by slip during SME. Introduction of dislocations influences on SME characteristics in the same way (see [51] for example). Apparently, in the case of thermal cycling of ZrCu alloy diffusional processes play an important role (unlike the TiNi alloy [50]) because of the wide temperature range of the thermal cycling (-196–400° C). It confirms indirectly by dividing of the MT into two stages during thermal cycling (Figs. 46, 47, 49) and directly by the metallographic and electron microscopic studies when the precipitation of the second phase on the grain boundaries were found.

### 2. NON-MARTENSITIC SHAPE MEMORY EFFECT DRIVEN BY AGEING PROCESSES

The first report about the non-martensitic shape memory effect called 'length memory effect' were published in [52]. Then, the authors of [53] studied the structural, phase composition and long time strength at 1173–1373 K of directionally solidified Ni-base eutectic COTAC-744 [54] with  $\gamma/\gamma'$ -MeC structure [53]. It was shown that the quenching from the temperature of full dissolution of the  $\gamma'$ -phase and ageing at 1173 K for 108 ksec provide the optimal size and morphology of  $\gamma'$ -phase that is responsible for the increase of the long time strength at 1173–1373 K in this material. Repeated thermal treatment (RTT) consisted in annealing at 1473 K during 3.6-9 ksec of the specimens subjected previously to creep tests on 1.5-1.8% results in general restoration of a)  $\gamma'$ -phase size and morphology, b) defect structure, c) mechanical properties, and d) geometrical dimensions (this could be the reason for the appearance of the 'length memory effect' term) similar to [52]. The authors of [53] described the retained elastic stresses in  $\gamma/\gamma'$ -matrix and single crystal MeC (Me-Nb, Ta) monocarbide fibres as the driving force of the effect mentioned above. These fibres are elongated elastically up to 1.5–1.8% of creep test deformation. At the same time, the matrix undergoing plastic deformation. After the creep test the fibres are deformed elastically by the elongation stresses and as a result the fibres applying the compression stresses on a matrix. After the RTT these compression stresses cause the restoration of the matrix shape and elastic unloading of the fibres. At the creep test deformations higher than 1.8% the relaxation of the retained stresses by the fibres fragmentation takes place that causes to the disappearance of the so-called 'length memory effect' [53].

So, it is clearly seen that the existence of the MT in material is enough for the SME but not necessary. The general necessary condition for the SME is the existence of the structural or microstructural reversibility caused by the non-relaxed (elastic) internal stresses appeared under the process of the accumulation of the deformation during the structural transformation.

Bublei, Koval, and Pasko [55] have examined shape recovery during heating in Cu–Al–Mn alloys after dynamic ageing in a state of strain.

They used ones containing 11–12 mass.% Al and from 2 to 6 mass.% Mn; they were melted in an induction furnace in argon and were then homogenized in the  $\beta$ -phase stability region (at 850°C) for 20 h. The compositions were determined by X-ray spectral means. The specimens were quenched in water from 900°C. The heat treatment (ageing up to 400°C) was in a salt bath.

The martensite transformation temperature ranges and the shape recovery characteristics (temperature dependence and degree of recovery) were determined according to [22]. The phase compositions were monitored by XRD with Cu $K_{\alpha}$  radiation.

The  $30 \times 2 \times 0.4$  mm plates with the composition Cu + 11.4 mass.% Al + + 3.4 mass.% Mn were given to dynamic ageing after quenching: they were bent in a cylindrical mount 20 mm in diameter to 2% strain and in that state were aged between 250 and 400°C.

In the quenched state, the specimens showed shape memory related to the martensite transformation (Fig. 50a); on ageing a specimen that had not undergone preliminary dynamic ageing, there were no marked changes in the temperature range and shape recovery on ageing for up to 30 min at 250–400°C. Similar results have been obtained [56, 57] for Cu–Al–Mn alloys similar in composition.

The shape memory effect in dynamically aged specimens was different from that in quenched ones. Figure 50b shows the bending change on



**Figure 50.** Variation in deflection  $\delta$  (arbitrary units) on heating of quenched alloy (a) and an alloy dynamically aged at 250°C for 5 min (b), composition Cu + 11.4 mass.% Al + 3.4 mass.% Mn.

heating for Cu + 11.4 mass. % Al + 3.4 mass. % Mn dynamically aged for 5 min at 250°C. The shape recovery is splitting in two stages: the first stage coincides with the temperature range for the reverse martensite transformation, while the second stage begins at 300–400°C. During subsequent cooling and heating in a dynamically aged specimen, there was no shape recovery at high temperatures (second stage). The shape recovery curve did not differ from that for the variation in bending on heating and cooling for the quenched alloy. The resistivity during martensite transformation was examined as a function of temperature, which showed no fall in the shape recovery temperature range 300–400°C, which shows that the high temperature recovery is not of martensitic type.

Ageing at 300–400°C for deformed specimens had different effect on the two stages. The low-temperature stage is unaffected with or without dynamic ageing. The shape recovery at high temperatures is reduced from 100% on ageing for 1 min at 250°C to 20% with 1 min at 400°C (Fig. 51). The range in which the high-temperature recovery occurs shifts upwards.

The quenched specimens were given various sequences of strain and ageing at 250°C; ones previously aged for 1 min at 250°C and then strained in the mount at room temperature showed no shape recovery at high temperatures. The high-temperature effect can occur only on dynamic ageing for slightly decomposed  $\beta_1$  phase.

XRD was applied to the changes on dynamic ageing; the phase diagrams for these alloys show that the  $\beta_1$  phase at 410°C contains Cu + 12 mass.% Al + 7.5 mass.% Mn and decomposes by a eutectoid mechanism with the formation of stable  $\alpha$ ,  $\gamma_2$  and Cu<sub>3</sub>Mn<sub>2</sub>Al phases. The XRD did not indicate any changes in structure in those phases in alloys after dynamic ageing. The guenched and dynamically aged specimens gave lines from the  $\alpha$ ,  $\gamma_2$  and  $\beta_1$ phases, and the last, as in other Cu–Al alloys, showed  $B1 + D0_3$  ordering. The lattice parameter for the  $\beta_1$  phase for quenched Cu + 11.4 mass.% Al + 3.4 mass.% Mn was a = 0.288 nm. Ageing at 250°C for up to 60 min increased monotonously to 0.292 nm, where there was some broadening for the  $\beta_1$  lines, which may be due to isostructural decomposition of the  $\beta_1$ phase with the deposition of Cu<sub>2</sub>MnAl particles in the Cu<sub>3</sub>Al matrix [58]. The amounts of the  $\alpha$ - and  $\gamma_2$ -phases were only 10–15% in the quenched and dynamically aged specimens that showed the high-temperature shape memory effect. The  $\gamma_2$  phase has an f.c.c. lattice of  $\gamma_2$  brass type and corresponds to Cu<sub>9</sub>Al<sub>4</sub> in binary alloys, while the  $\alpha$ -phase is a solid solution of manganese and aluminium in copper.

Increasing the dynamic-ageing time or raising the ageing temperature increases the amounts of  $\alpha$  and  $\gamma_2$  phases segregating; at the same time, there is a reduction in the deflection on shape recovery at 250–350°C, with the range in which the effect occurs shifting to higher temperatures (Fig. 52).

The quenched and dynamically aged specimens had the same phase composition, so the high-temperature effect may be due to the initial



**Figure 51.** Temperature for the high temperature shape memory effect ( $T_{ef}$ ) (1) and shape recovery ratio  $\delta/\delta_{in}$  in % (2) as a function of dynamic ageing temperature ( $T_a$ ) (t—1 min).



**Figure 52.** Temperature for the high temperature shape memory effect ( $T_{ef}$ ) (1) and shape recovery ratio  $\delta/\delta_{in}$  in % (2) as a function of dynamic ageing time (*t*) ( $T_{a}$ —350°C)

stages of eutectoid decomposition in the  $\beta_1$  phase, where the amounts of the segregating stable phases do not exceed 10%, namely one gets  $\alpha$ and  $\gamma_2$  particles coherent with the  $\beta_1$  matrix. Coherency is lost at a size  $D_{cr}$ and then the high-temperature shape memory effect weakens. At that time, which one can identify from hardness increase in the  $\beta_1$  phase [59], there is no high-temperature shape memory effect.

Two-stage shape memory has been observed in Cu–Al–Mn alloys given dynamic ageing in the strained state after quenching; the first stage involves a martensite mechanism, while the second is not of martensite type. The range and extent of the recovery for the latter are dependent on the ageing temperature. That stage may be related to the initial steps in the eutectoid decomposition of the  $\beta_1$  phase on ageing.

Another example of the similar behaviour can be found in [60]. The SME in Ti–52 at.% Ni alloy was found in bending conditions at 773–973 K. The authors of this paper attributed this effect to the precipitation-dissolution processes of the  $Ti_3Ni_4$  coherent particles. They also explained the process of the accumulation of the deformation in the terms of anisotropic mismatch between matrix and particles in addition to the preferential formation of the differently oriented particles variants in compressed and elongated sides of the specimens under constraint ageing in bending. Unfortunately, they did not show any structural result confirming their model.

Koval, Monastyrski and Firstov have studied the precipitation processes in Fe–Ni–Co–Ti system and obtained the same effect in the same temperature range [61].

The alloy studied in this investigation had the following composition: Fe-



**Figure 53.** Shape memory effect in Fe–34.9 at.% Ni–13.8 at.% Co–4.4 at.% Ti alloy after preliminary quenching from 1473 K ( $K_{SME} = 30-40\%$ ).

34.9 at.% Ni–13.8 at.% Co–4.4 at.% Ti. The specimens quenched from 1473 K into water were tested using bending technique [22] in the temperature range 300–1100 K under the static load with the 1–2 K/sec heating rate. After the loading at room temperature and heating the shape change was not observed up to 700 K (Fig. 53). The subsequent heating was accompanied by the bending of the specimens. Then, at 1000 K the specimens were quickly cooled (50–100 K/sec) and released from the static load. Repeated heating exhibited the shape recovery in the 700–900 K temperature range. The shape recovery ratio in this case was equal to 30–40%. It was suggested that this effect caused by the precipitation processes of  $\gamma'$ -phase (Ni<sub>3</sub>Ti) that takes place in the same temperature range.

Electron microscopic investigation of the quenched specimen subjected to the first heating under static load confirmed our suggestion (Fig. 54). In the bright (Fig. 54a) and dark fields (Fig. 54b) the fine particles (5 nm) of the  $\gamma'$ -phase preferentially formed on the dislocation nets are clearly seen.

The dislocations density in deformed specimens was higher than in quenched ones. So, it was concluded that the incomplete shape recovery (Fig. 53) was caused by plastic deformation processes. That is why we have studied the hardening of the matrix by means of  $\gamma'$ -phase precipita-



**Figure 54.** Structure of the Fe–34.9 at.% Ni–13.8 at.% Co–4.4 at.% Ti alloy after preliminary quenching from 1473 K (elongated region of the specimen accumulated deformation during heating in bending): bright field (a), dark field taken in (001) superstructural reflection (b), diffraction pattern, zone axis [211] (c).

tion during ageing at 973 K. The results of microhardness measurements are shown in Fig. 55. The significant increase of hardness is seen at the nucleation and growth stage. Then, hardness passes through a maximum at 400–500 seconds of ageing and increases up to ~ 1500 MPa at 1 h of ageing. Such strengthening of the matrix is quite enough for suppressing of plastic deformation. Therefore, this time of ageing was chosen for the improving of the incomplete SME shown in Fig. 53.

The SME behaviour of the preliminary aged samples is shown in Fig. 56. In this case the nearly complete shape recovery was found. The subsequent cycling did not change the shape recovery ratio but the temperature range of bending and shape recovery processes was shifted to higher temperatures.

Electron microscopic investigations of the preliminary aged samples after bending were performed on specimens cut from the both sides (elongated (Fig. 57) and compressed (Fig. 58)). The drastic difference in the dimensions of the particles can be seen. In elongated region the diameter of particles is ~20 nm and in compressed ~10 nm. No visible at-



**Figure 55.** Microhardness changes *vs* ageing time at 973 K for Fe–34.9 at.% Ni– 13.8 at.% Co–4.4 at.% Ti alloy after preliminary quenching from 1473 K.

tributes of plastic deformation processes can be seen. After the restoration of the original shape the size of the particles on both sides equalizes and becomes equal to  $\sim 5$  nm (Fig. 59).

It can thus be supposed that in the case of the non-martensitic SME in ausaged Fe–34.9 at.% Ni–13.8 at.% Co–4.4 at.% Ti alloy the difference in size of the spherical  $\gamma'$ -phase particles on the different sides of the bent specimens is the origin of the high temperature deformation. Equalizing of



**Figure 56.** Shape memory effect in Fe–34.9 at.% Ni–13.8 at.% Co–4.4 at.% Ti alloy after preliminary quenching from 1473 K and subsequent ageing at 973 K during 1 h ( $K_{\text{SME}} = 90-100\%$ ).



**Figure 57.** Structure of the Fe–34.9 at.% Ni–13.8 at.% Co–4.4 at.% Ti alloy after preliminary quenching from 1473 K and subsequent ageing at 973 K during 1 h (elongated region of the specimen accumulated deformation during heating in bending): dark field taken in (001) superstructural reflection (a), diffraction pattern, zone axis [310] (b).



**Figure 58.** Structure of the Fe–34.9 at.% Ni–13.8 at.% Co–4.4 at.% Ti alloy after preliminary quenching from 1473 K and subsequent ageing at 973 K during 1 h (compressed region of the specimen accumulated deformation during heating in bending): dark field taken in (001) superstructural reflection (a), diffraction pattern, zone axis [210] (b).

that sizes after the nearly complete shape recovery confirms this supposition and leads us to agree with the conclusions made in Ref. [55].

Really, in our case also the precipitation-dissolution processes of the second phase play the major role in the non-martensitic SME. As for the process of accumulation of the deformation, we can not completely agree with the authors of [55] because they exploit the plate-like shape of the particles in their model. In our case the particles are spherical but the result is the same. Anyway, it is too early to give the definite answer on the mechanism of this process.



**Figure 59.** Structure of the Fe–34.9 at.% Ni–13.8 at.% Co–4.4 at.% Ti alloy after preliminary quenching from 1473 K and subsequent ageing at 973 K during 1 h (after the shape recovery; both sides of the specimen have the same structural state): dark field taken in (001) superstructural reflection (a), diffraction pattern, zone axis [110] (b).

Additional quantitative measurements of the size, density and distribution of the particles in different constraint conditions will help us in better understanding of the mechanism of the non-martensitic SME.

Naturally, this effect is far from the industrial application now but the high temperatures of the precipitation for some materials still leave us the possibility to use it in one-way shape memory thermal sensitive elements when the other SMA are out of use.

#### **3. CONCLUSIONS**

High-temperature *B*2 phase of Zr-based equiatomic intermetallics transforms martensitically on cooling mostly into two monotonic martensites.

Interaction between the thin plates of these two types of martensite results in the in the wide temperature hysteresis, prolonged temperature intervals of forward and reverse MT's and large amount of energy dissipated during full cycle of MT in Zr-based equiatomic intermetallics in contrast to materials undergoing MT of thermoelastic type. MT in *B2* compounds of Zr is accompanied by the perfect SME even at high temperatures in spite of wide temperature hysteresis. Good shape memory properties, high oxidation resistance up to 800 K and comparative cheapness (excluding ZrRh and ZrIr) make the quasi-binary intermetallics of Zr competitive with existing SMA's in the possible high-temperature applications.

Non-martensitic SME driven by the precipitation-dissolution processes of second phase in constrained conditions is far from the industrial application now but the high temperatures of the precipitation for some materials still leave us the possibility to use it in one-way shape memory thermal sensitive elements when the other SMA's are out of use.

#### REFERENCES

- 1. E. M. Carvalho and I. R. Harris, J. Less-Common Met., 106: 129 (1985).
- 2. Yu. N. Koval, G. S. Firstov, and A. V. Kotko, Scripta Met. Mat., 27: 1611 (1992).
- 3. V. V. Nemoshkalenko, A. V. Zhalko-Titarenko, Yu. N. Koval et al., *Metallofiz. Noveishie Technol.*, **15**, No. 1: 12 (1993) (in Russian).
- 4. F. E. Wang, J. Appl. Phys., 38: 822 (1967).
- 5. F. E. Wang and W. Ernst, J. Appl. Phys., 39: 2192 (1968).
- 6. I. R. Harris, D. Hossain, and K. G. Barraclough, *Scripta Met.*, **4**: 305 (1970).
- 7. D. Hossain, I. R. Harris, and K. G. Barraclough, *J. Less-Common Met.*, **37**: 35 (1974).
- 8. C. Lall, M. H. Loretto, and I. R. Harris, *Acta Metall.*, **26**: 1631 (1978).
- 9. E. M. Carvalho and I. R. Harris, J. Less-Common Met., 106: 117 (1985).
- 10. E. M. Carvalho and I. R. Harris, J. Less-Common Met., 106: 143 (1985).
- 11. E. M. Carvalho and I. R. Harris, J. Mat. Sci., 15: 1224 (1980).
- 12. I. Z. Jorda, T. Graf, and L. Schellenberg, *J. Less-Common Met.*, **136**, No. 1: 313 (1988).
- E. L. Semenova and Yu. V. Kudryavtsev, J. Alloys and Compounds, 203: 165 (1994).
- Yu. N. Koval, G. S. Firstov, L. Delaey, and J. Van Humbeeck, *Scripta Met. et Mat.*, 31, No. 7: 799 (1994).
- W. Y. Jang, J. Van Humbeeck, L. Delaey et al., 'The Influence of Ti and Ni Additions and Thermal Cycling on the Martensitic Transformation in CuZr Alloys' (Proc. of the Advanced Materials'93, B: Shape Memory Materials and Hydrides) (Eds. K. Otsuka et al.), Trans. Mat. Res. Soc. Jpn., **18B** (Elsevier: 1994), p. 1009.
- 16. A. V. Zhalko-Titarenko, M. L. Yevlashina, V. N. Antonov et al., *Phys. Stat. Sol. (b)*, **184**: 121 (1994).
- Yu. N. Koval and G. S. Firstov, 'Crystal Structure of the ZrCu Intermetallic Compound' (Proc. of XVIth Conference on Applied Crystallography) (Cieszyn, Poland, 22–26 August, 1994) (Eds. H. Morawiec and D. Stróz) (Singapore: World Scientific: 1995), p. 310.
- 18. W. Bührer, R. Gotthardt, A. Kulik et al., J. Phys. F: Met. Phys., 13: L77 (1983).
- 19. Y. Kudoh, M. Tokonami, S. Miyazaki, and K. Otsuka, *Acta Metall.*, **33**, No. 11: 2049 (1985).
- R. M. Waterstrat, L. A. Bendersky, and R. Kuentzler, 'Deformation Twins and Martensite in Ductile B2 Alloys of the Zr(RuPd) System' (Proc. of the ICOMAT-92) (Monterey, USA, 20–24 July, 1992) (Ed. J. Perkins) (Monterey Institute for Advanced Studies: 1993), p. 545.
- 21. I. Groma, J. Lendvai, A. Cziráki et al., Scripta Met. et Mat., 26: 255 (1992).
- 22. V. V. Martynov and L. G. Khandros, *Fiz. Met. Metalloved.*, **39**, No. 5: 1037 (1975) (in Russian).
- 23. D. B. Wiles and R. A. Young, J. Appl. Cryst., 14: 149 (1981).
- 24. D. Schryvers, G. S. Firstov, J. W. Seo et al., Scripta Mat., 36: 1119 (1997).
- K. Shimizu and K. Otsuka, Shape Memory Effects in Alloys (Ed. J. Perkins) (New York: Plenum Press: 1975), p. 59–83.
- 26. J. Ortin and A. Planes, Acta Metall., 36, No. 8: 1873 (1988).
- 27. A. Cornelis and C. M. Wayman, Scripta Met., 10: 359 (1976).
- D. S. Lieberman, M. A. Schmerling, and R. S. Karz, Shape Memory Effects in Alloys (Ed. J. Perkins) (New York: Plenum Press: 1975), p. 203–226.

- 29. B. Predel and W. Schwerman, Z. Naturforschg., 22a: 1499 (1967).
- G. Airoldi, B. Rivolta, and C. Turco, 'Heats of Transformations as a Function of Thermal Cycling in NiTi Alloys' (Proc. of the ICOMAT-86) (Nara, Japan, 26–30 August, 1986) (Sendai: The Japan Institute of Metals: 1987), p. 691–696.
- 31. Y. C. Lo, S. K. Wu, and C. M. Wayman, Scripta Met. et Mat., 24: 1571 (1990).
- Yu. N. Koval, G. S. Firstov, J. Van Humbeeck et al., 'B2 Intermetallic Compounds of Zr. New Class of the Shape Memory Alloys' (Proc. of the ICOMAT-95) (Lausanne,Switzerland, 20–25 August, 1995) (Eds. R. Gotthardt and J. Van Humbeeck), J. Physique IV, C8, 5 (Les Editions de Physique, 1995), p. 1103–1108.
- G. V. Kurdjumov and L. G. Khandros, *Dokl. Akad. Nauk SSSR*, 66: 211 (1949) (in Russian).
- 34. G. S. Firstov, Yu. N. Koval, and J. Van Humbeeck (unpublished results).
- 35. R. A. Young and A. Sakthivel, *Program DBWS-9006PC for Rietveld Analysis of X-Ray and Neutron Powder Diffraction Patterns* (release of 1991).
- 36. R. J. Hill and C. J. Howard, *J. Appl. Cryst.*, **20**: 467 (1987).
- G. S. Firstov, Yu. N. Koval, and J. Van Humbeeck, 'Irreversible Processes During Martensitic Transformation in Zr-Based Shape Memory Alloys' (Proc. of the ESOMAT'97) (Enschede, The Netherlands, 1–5 July, 1997) (Eds. J. Beyer, A. Bu ttger, and J. H. Mulder), J. Physique IV, C5, 7: 549 (1997).
- G. S. Firstov, Yu. N. Koval, and J. Van Humbeeck, 'Shape Memory Behaviour in Quasi-Binary ZrCu-Based Intermetallic Compounds' (Proc. of the SMST'99) (Antwerp Zoo, Belgium, 5–9 September, 1999) (to be published).
- 39. G. Roebben, B. Bollen, A. Brebels et al., Rev. Sci. Instrum., 68: 4511 (1997).
- 40. A. S. Nowick and B. S. Berry, in *Anelastic Relaxation in Crystalline Solids* (New York: Academic Press: 1972).
- 41. D. B. Wiles and R. A. Young, J. Appl. Cryst., 14: 149 (1981).
- 42. J. W. Seo and D. Schryvers, Acta mater., 46, No. 4: 1165 (1998).
- 43. S. A. Shabalovskaya, *Solid State Com.*, **70**, No. 1: 23 (1989).
- 44. V. V. Kokorin, I. A. Osipenko, and T. V. Shirina, *Fiz. Met. Metalloved.*, **67**, No. 3: 601 (1989) (in Russian).
- S. A. Shabalovskaya, 'Factors Determining B2 Phase Stability in Ti-Based Compounds' (Proc. of the ICOMAT-92) (Monterey, USA, 20–24 July, 1992) (Eds. C. M. Wayman and J. Perkins) (Monterey Institute for Advanced Studies: 1993), p. 123.
- A. M. Kostyshin and G. A. Takzey, *Dynamic Magnetic Susceptibility of Spin Glasses*' (Kiev: 1985) (Prepr./Academy of Sciences of the Ukr.SSR. Institute for Metal Physics. 4.85).
- 47. V. N. German, A. A. Bakanova, L. A. Tarasova, and Yu. N. Sumulov, *Soviet Physics–Solid State* (Translated from *Fizika Tverdogo Tela*), **12**, No. 2: 490 (1970).
- 48. E. C. Stoner, Proc. R. Soc. London, Ser. A, 154: 656 (1936).
- 49. A. N. Das and B. Ghosh, J. Phys. C: Solid State Phys., 16: 1799 (1983).
- 50. S. Miyazaki, Y. Igo, and K. Otsuka, Acta Met., 34, No. 10: 2045 (1986).
- 51. Yu. N. Koval, A. P. Kozlov, and G. E. Monastyrski, *Scripta Met.*, **23**, No. 10: 1731 (1989).
- T. Khan, G. F. Stohr, and H. Bibring, On Optimized DS Composite for Turbine Blades (Proc. of the Fourth International Symposium on Superalloys) (ASM: 1980), p. 531–540.
- 53. S. T. Kishkin, I. L. Svetlov, L. P. Sorokina et al., *Fiz. Met. Metalloved.*, **55**, No. 4: 754 (1983) (in Russian).
- 54. H. Bibring, T. Khan, M. Rabinovich, and G. F. Stohr, *Development and Evaluation* of New Industrial DS Monocarbide Reinforced Composites for Turbine Blades

354

(*Proc. of the Third International Symposium on Superalloys*) (Seven Springs, Pennsylvania, USA: 1976), p. 331–345.

- 55. I. R. Bublei, Yu. N. Koval, and A. Yu. Pasko, *Phys. Metals*, **10**, No. 5: 902 (1991).
- 56. L. G. Khandros and I. A. Arbuzova, *Metally, Ehlektrony, Reshyotka* (Metals, Electrons, and Lattice) (Kiev: Naukova Dumka: 1975), p. 109–143 (in Russian).
- 57. T. L. Dobrovol'skaya, P. V. Titov, and L. G. Khandros, *Martensitic Transformation in Cu–Al–Mn Alloys After Partial Decomposition of the β1 Phase (Proc. of the ICOMAT-77)* (Kiev, USSR, 16–20 May, 1977) (Eds. V. N. Gridnev and L. G. Khandros); *Martensitnyye Prevrashcheniya v Metallakh i Splavakh* (Martensitic Transformations in Metals and Alloys) (Kiev: Naukova Dumka: 1979), p. 155–157 (in Russian).
- 58. M. Bouchard and G. Thomas, Acta Met., 23, No. 12: 1485 (1975).
- 59. A. Kelly and R. Nicholson, *Dispersion Hardening* (Moscow: Mir: 1968) (Russian translation).
- 60. R. Kainuma and M. Matsumoto, Scripta Met., 22: 475 (1988).
- Yu. N. Koval, G. E. Monastyrski, and G. S. Firstov, *Method of Fabrication of the Thermal Sensitive Element—Temperature Transducer* (Inventor's Certificate of the USSR No. 1469901), *Byulleten' Izobretenij i Otkrytij* (Bulletin of the Inventions and Discoveries), No. 12, p. 244 (1989).